

Highlight on Direct Amination of Unreactive C–H Bonds Catalyzed by *N*-hydroxyphthalimide

Dr. Kexian Chen ^{[a],[b]}, Dr. Pengfei Zhang ^{[a],[c]} and Prof. Haoran Li ^{[a],[b]*}

^[a] Department of Chemistry, Zhejiang University, Hangzhou 310027, P.R. China. E-mail: lihr@zju.edu.cn

^[b] State Key Laboratory of Chemical Engineering, Department of Chemical and Biological Engineering, Zhejiang University, Hangzhou 310027, P.R. China

^[c] Chemical Sciences Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, United States

The selectively direct heteroatom-functionalization of unreactive C–H bonds of readily available organic substrates (e.g. methane [1]) or complex macromolecules *via* simplified synthetic routes under mild conditions is of crucial importance in the academic and industrial areas, which is a research hotspot at present and in future. In particular, economic-friendly standards [2] require reagents, products, and conditions in such reactions to be capable of low environmental impact, namely green enough. Compared to conventional C–H transformation protocols, recently developed innovative *N*-hydroxyphthalimide (NHPI)-based catalytic systems by Ishii *et al.* [3–5] received great attentions [3–8] due to its environmental benign character and established desired high reactivity.

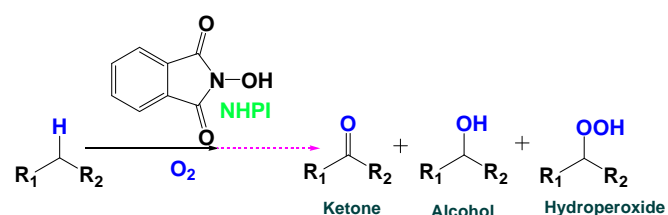


Figure 1. Aerobic oxidation of hydrocarbons under mild conditions catalyzed by *N*-hydroxyphthalimide (NHPI) in the presence or absence of any co-catalysts.

Generally, NHPI is mainly used in the aerobic oxidation reactions. Serving as an efficient carbon-radical-producing catalyst [4], its catalytic reactivity is induced by its *in situ* generated phthalimide *N*-oxyl radical (PINO) [9] via C–H activation; then the resultant carbon radicals combine the dioxygen to produce oxygenated compounds like alcohols, ketones, hydroperoxides, and carboxylic acids (Figure 1). Its utilization has

covered the oxidation of a variety of hydrocarbons, including cholesterols [10,11], isophorones [12,13], cyclohexene [14] and toluenes [15]. The analogous catalysts [9,14,16,17] have also been designed, developed, and investigated for such purpose due to their similar catalytic features.

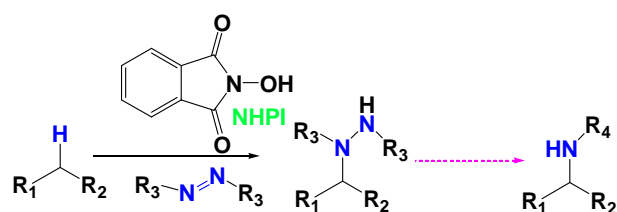


Figure 2. Radical amination of various C–H bonds by the combination of *N*-hydroxyphthalimide (NHPI) and dialkyl azodicarboxylate, reported by Inoue, *et al.* [18].

Recently, Japanese scientist Inoue and his cooperators [18] developed a novel method for direct radical amination of benzylic, propargylic, and aliphatic C(sp³)–H bonds catalyzed by NHPI in combination of stoichiometric dialkyl azodicarboxylate (Figure 2), which is quite different from the above general reactions. C–N construction is very appealing as the occurrence of such bonds in nitrogen-substituted natural and non-natural biologically active molecules, e.g. alkaloids and potent pharmaceutical drugs [19], is quite universal. Such construction from unreactive C–H bonds requires predictable chemo-, regio-, and stereo-selective characters, making it one of the most challenging issues in organic synthesis. The fact that the traditional strategies through nitrene insertion commonly utilize transition metal catalysts [20] makes the exploitation of such catalytic reactions in a green manner urgent and

vitaly important. The novel metal-free strategy through radical routes developed by Inoue *et al.* [18] shows green features and high yields of nitrogen-containing products for a variety of organic substrates, which sheds significant light on the state-of-the-art C-H amination, and will certainly give impetus to witness the feasibility of more efficient and sustainable C-N construction for natural products and pharmaceuticals under environmental benign conditions in future. It can also be anticipated from their studies that some of NHPI derivatives or analogs [9,14,16,17] can serve as appropriate catalysts for these or other transformations (e.g. C-F [20]), which still needs experimental evidence to make such a conclusion.

References:

- [1] X. Hu, *et al.*, *Theoretical study on the structure-reactivity relationships of acetylacetonate-Fe catalyst modified by ionic compound in C-H activation reaction*. *J. Catal.* 2010, **272** (2): p. 320-332.
- [2] P.J. Dunn, *The importance of green chemistry in process research and development*. *Chem. Soc. Rev.* 2012, **41** (4): p. 1452-1461.
- [3] Y. Ishii, *et al.*, *A novel catalysis of N-hydroxyphthalimide in the oxidation of organic substrates by molecular oxygen*. *J. Org. Chem.* 1995, **60** (13): p. 3934-3935.
- [4] Y. Ishii, *et al.*, *Innovation of hydrocarbon oxidation with molecular oxygen and related reactions*. *Adv. Synth. Catal.* 2001, **343** (5): p. 393-427.
- [5] Y. Ishii and S. Sakaguchi, *Recent progress in aerobic oxidation of hydrocarbons by N-hydroxyimides*. *Catal. Today* 2006, **117** (1-3): p. 105-113.
- [6] R.A. Sheldon and I.W.C.E. Arends, *Organocatalytic oxidations mediated by nitroxyl radicals*. *Adv. Synth. Catal.* 2004, **346** (9-10): p. 1051-1071.
- [7] F. Recupero and C. Punta, *Free radical functionalization of organic compounds catalyzed by N-hydroxyphthalimide*. *Chem. Rev.* 2007, **107** (9): p. 3800-3842.
- [8] S. Coseri, *Phthalimide-N-oxyl (PINO) radical, a powerful catalytic agent: Its generation and versatility towards various organic substrates*. *Catal. Rev.* 2009, **51** (2): p. 218-292.
- [9] K. Chen, *et al.*, *Theoretical studies on multi-hydroxyimides as highly efficient catalysts for aerobic oxidation*. *ChemPhysChem* 2013, **14** (1): p. 179-184. The *in situ* PINO generated methods are briefly summarized in this paper.
- [10] Z. Yao, *et al.*, *An environmentally benign catalytic oxidation of cholesterol with molecular oxygen by using N-hydroxyphthalimide*. *Green Chem.* 2009, **11** (12): p. 2013-2017.
- [11] P. Zhang, *et al.*, *Acetylacetonate-metal catalyst modified by pyridinium salt group applied to the NHPI-catalyzed oxidation of cholesteryl acetate*. *Catal. Sci. Tech.* 2011, **1** (7): p. 1133-1137.
- [12] C. Wang, *et al.*, *Metal and solvent-free oxidation of α -isophorone to ketoisophorone by molecular oxygen*. *Catal. Commun.* 2010, **11** (8): p. 758-762.
- [13] K. Chen, *et al.*, *Aerobic oxidation of β -isophorone catalyzed by N-hydroxyphthalimide: The key features and mechanism elucidated*. *Phys. Chem. Chem. Phys.* 2012, **14** (35): p. 12141-12146.
- [14] Y. Sun, *et al.*, *Correlation analysis of the substituent electronic effects on the allylic H-abstraction in cyclohexene by phthalimide-N-oxyl radicals: A DFT study*. *J. Phys. Chem. B* 2010, **114** (14): p. 4862-4869.
- [15] I. Hermans, *et al.*, *Pronounced non-Arrhenius behaviour of hydrogen-abstractions from toluene and derivatives by phthalimide-N-oxyl radicals: A theoretical study*. *Phys. Chem. Chem. Phys.* 2008, **10** (8): p. 1125-1132.
- [16] I. Hermans, *et al.*, *Autoxidation catalysis with N-hydroxyimides: more-reactive radicals or just more radicals?* *Phys. Chem. Chem. Phys.* 2007, **9** (6): p. 686-690.
- [17] K. Chen, *Design and Mechanistic Investigations of N-Hydroxyimides as Organocatalysts for Aerobic Oxidation*. PhD Thesis, Zhejiang Univ., 2009.
- [18] Y. Amaoka, *et al.*, *Radical amination of C(sp³)-H bonds using N-hydroxyphthalimide and dialkyl azodicarboxylate*. *J. Org. Chem.* 2012, **77** (22): p. 9959-9969.
- [19] See recent reviews and selected reports in H. Chen, *et al.*, *Copper-catalyzed aliphatic C-H amination with an amidine moiety*. *Org. Lett.* 2013, **15** (1): p. 212-215.
- [20] Y. Amaoka, *et al.*, *Metal-free fluorination of C(sp³)-H bonds using a catalytic N-oxyl radical*. *Org. Lett.* 2013, **15** (9): p. 2160-2163.